

No. of Printed Pages : 5

**MCH-016**

**M. SC. IN CHEMISTRY/  
M. SC. IN ANALYTICAL CHEMISTRY  
(MSCCHEM/MSCANCHEM)**

**Term-End Examination**

**December, 2025**

**MCH-016 : INORGANIC CHEMISTRY-II**

*Time : 2 Hours*

*Maximum Marks : 50*

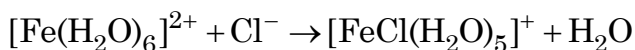
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**Note :** Answer any *five* questions. All questions carry equal marks.

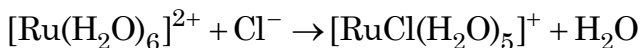
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1. (a) Name the *four* classes of substitution reactions in inorganic chemistry. What is the basis of this classification?      5

- (b) What are the different classes of metal ions depending on their substitution rates ? Based on this, give reasons for the different rate constants for the following two reactions : 5



$$k = 10^6 \text{ m}^{-1} \text{ s}^{-1}$$



$$k = 10^{-2} \text{ m}^{-1} \text{ s}^{-1}$$

2. (a) Explain trans-effect in ligand substitution in square planar complexes with the example of platinum complexes. Also give the energy coordinate diagrams showing poor

trans-effect and  $\pi$ -bonding effect (lower transition state). 6

(b) Give the classification of ligands based on their trans-effects with the help of suitable examples. 4

3. (a) Draw the figures to depict  $\text{cis } \wedge$  and  $\text{cis } \Delta$  for a six-coordinate octahedral complex. Which axis of symmetry is present in them ? 5

(b) Give reasons for the faster outer-sphere reactions for 2nd and 3rd row transition metal ions compared to the 1st row transition metal ions. 5

4. (a) With the help of suitable examples, explain two electron transfer reactions.

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- (b) Give any *five* salient features of mixed valence coordination complexes. 5
5. (a) What are siderophores ? When are they called siderochromes ? Give any *two* uses of siderophores. 5
- (b) What are rubredoxins ? Give an account of the different types of ferredoxins. 5
6. (a) Which metal ion is present in chlorophyll ? Why is it suitable and why heavy transition metal ions are not suitable for the plant species ? 5
- (b) Explain the structure of carboxypeptidase A. Which metal is

present in it and why is it suitable as a catalyst in biological systems ? 5

7. (a) Give an account of the redox chemistry of metals with nucleic acids. 5
- (b) What is the effect of thallium poisoning in humans ? Which enzymes can it inhibit and how can it be dealt with ? 5

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